## **PhytoTechnology Laboratories®**



Helping to Build a Better Tomorrow through Plant Science™

## **Product Information Sheet**

E582

EDTA Solution 0.5 M, pH 8.0

Synonyms: N/A

CAS: 6381-92-6

Formula: C<sub>10</sub>H<sub>14</sub>N<sub>2</sub>O<sub>8</sub>Na<sub>2</sub>•2H<sub>2</sub>O

Mol. Weight: 372.24

**Properties** 

Form: Liquid

Appearance: Clear, Colorless Solubility: Miscible with Water

Application: Molecular Biology, Cell Culture

Storage Temp: Room Temperature

Typical Working Concentration:

Varies according to application

Notes: Sterile filte

Sterile filtered, RNAse & DNAse Free

pH buffered to pH 8.0

NaO - C - CH<sub>2</sub> N - CH<sub>2</sub> CH<sub>2</sub> - N CH<sub>2</sub> - C - ONE

## **Application Notes**

EDTA is useful as a chelating agent due to its ability to sequester metal ions, like Ca<sub>2</sub><sup>+</sup> and Fe<sub>3</sub><sup>+</sup>. Some cations are responsible for various undesirable effects in cell culture processes. These metal ions will remain in the solution after being bound by EDTA, but exhibit a lowered activity. The use of EDTA therefore allows for greater control over many experiments in which an extent of cation control is necessary.

EDTA is also commonly used in electrophoresis buffer systems.

This solution is buffered using NaOH to reach a pH of 8.0.

## References

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